

## 4-Acetylanilinium chloride monohydrate

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## Key indicators

Single-crystal X-ray study

$T = 293\text{ K}$

Mean  $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$

$R$  factor = 0.032

$wR$  factor = 0.070

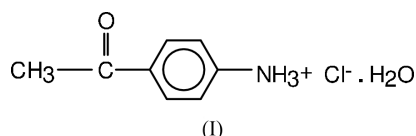
Data-to-parameter ratio = 18.0

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The crystal structure of the title compound,  $\text{C}_8\text{H}_{10}\text{NO}^+\cdot\text{Cl}^-\cdot\text{H}_2\text{O}$ , is composed of 4-acetylanilinium cations, chloride anions and water molecules of crystallization. The crystal packing features stacking interactions between the aromatic rings, with stacks running along the  $c$  axis of the crystal, as well as extensive  $\text{N}-\text{H}\cdots\text{O}$ ,  $\text{N}-\text{H}\cdots\text{Cl}$  and  $\text{O}-\text{H}\cdots\text{Cl}$  hydrogen bonding which involves all five 'active' H atoms and links cations, anions and water molecules into a three-dimensional framework.

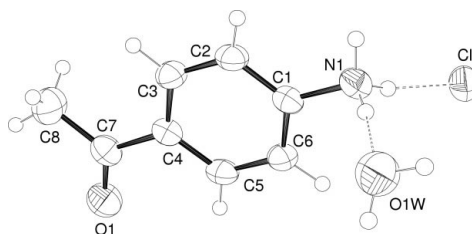
## Comment

The title salt, (I), is composed of 4-acetylanilinium ( $\text{C}_8\text{H}_{10}\text{NO}^+$ ) cations, chloride anions and water molecules of crystallization. The cation is a protonated aniline derivative (Fig. 1); protonation results in a lengthening of the C1–N1 bond to 1.454 (2) Å. This bond is significantly longer than that in non-protonated anilines [1.335 (3) Å (Goswami *et al.*, 1999), 1.386 (4) Å (Ploug-Sørensen & Andersen, 1985) and 1.391 (3) Å (Ploug-Sørensen & Andersen, 1986)] and is typical for anilinium cations [1.464 (2) Å (Ploug-Sørensen & Andersen, 1985) and 1.460 (2) Å (Ploug-Sørensen & Andersen, 1986)].



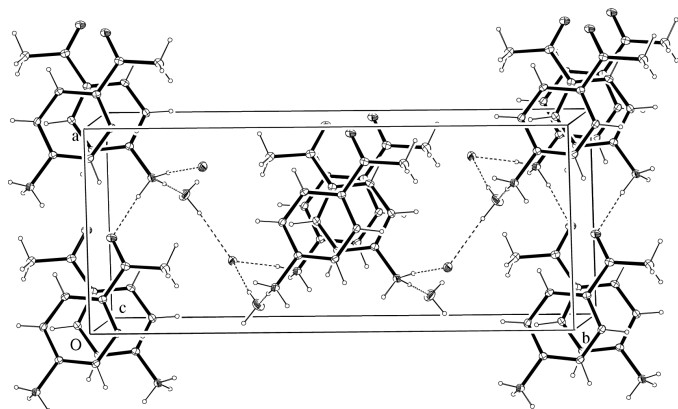
The most significant feature of the crystal packing is the stack of aromatic rings running along the  $c$  axis of the crystal; the distances between the centroids of neighbouring rings within the stack are 3.647 (2) and 3.648 (2) Å (Fig. 2). The chloride anions occupy the channels between the stacks.

All five 'active' H atoms in the structure are involved in hydrogen bonds of the  $\text{N}-\text{H}\cdots\text{O}$ ,  $\text{O}-\text{H}\cdots\text{Cl}$  and  $\text{N}-\text{H}\cdots\text{Cl}$  types (Table 2), linking all entities present in the structure into an infinite three-dimensional framework.



**Figure 1**

An ORTEP-3 (Farrugia, 1997) view of the cation, anion and water molecule in the structure of the title compound, showing the atom-labelling scheme and displacement ellipsoids at the 50% probability level. Hydrogen bonds are shown as dashed lines.



**Figure 2**  
View of the packing of the title compound. Hydrogen bonds are shown as dashed lines.

## Experimental

The title compound, (I), was prepared by bubbling gaseous hydrogen chloride through a solution of 4-acetylaniline (2 g) in ether (50 ml) for 15 min at an approximate flow rate of 80 ml min<sup>-1</sup>. The precipitate was filtered off and recrystallized from a mixture of ethanol and concentrated hydrochloric acid. The recrystallized product was dried in a desiccator, first over potassium hydroxide and then over phosphorus pentoxide.

### Crystal data

C<sub>8</sub>H<sub>10</sub>NO<sup>+</sup>·Cl<sup>-</sup>·H<sub>2</sub>O  
*M<sub>r</sub>* = 189.64  
 Monoclinic, *Cc*  
*a* = 7.690 (1) Å  
*b* = 17.930 (1) Å  
*c* = 7.095 (1) Å  
 $\beta$  = 95.70 (1)°  
*V* = 973.4 (2) Å<sup>3</sup>  
*Z* = 4

*D<sub>x</sub>* = 1.294 Mg m<sup>-3</sup>  
 Mo *K*α radiation  
 Cell parameters from 9241 reflections  
 $\theta$  = 2.3–28.9°  
 $\mu$  = 0.35 mm<sup>-1</sup>  
*T* = 293 (2) K  
 Prism, light yellow  
 0.30 × 0.21 × 0.21 mm

### Data collection

Stoe IPDS-II diffractometer  
 $\omega$  scans  
 Absorption correction: numerical  
 (*X-RED32*; Stoe & Cie, 2002)  
*T<sub>min</sub>* = 0.915, *T<sub>max</sub>* = 0.928  
 7188 measured reflections  
 2447 independent reflections

2060 reflections with *I* > 2σ(*I*)  
*R<sub>int</sub>* = 0.051  
 $\theta_{\max}$  = 28.9°  
*h* = -10 → 10  
*k* = -24 → 24  
*l* = -9 → 9

### Refinement

Refinement on *F*<sup>2</sup>  
*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.032  
*wR* (*F*<sup>2</sup>) = 0.070  
*S* = 0.96  
 2447 reflections  
 136 parameters  
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0444P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 ( $\Delta/\sigma$ )<sub>max</sub> = 0.001  
 $\Delta\rho_{\max} = 0.26 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.28 \text{ e \AA}^{-3}$   
 Absolute structure: Flack (1983),  
 1162 Friedel pairs  
 Flack parameter = 0.3 (2)

**Table 1**

Selected geometric parameters (Å, °).

C1–N1	1.454 (2)	C7–O1	1.211 (2)
C2–C1–N1	119.4 (1)	O1–C7–C8	121.5 (2)
C6–C1–N1	118.74 (13)	C4–C7–C8	118.8 (2)
O1–C7–C4	119.7 (1)		

**Table 2**

Hydrogen-bonding geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
N1–H1A···O1 <sup>i</sup>	0.89	1.94	2.826 (2)	175
O1W–H1W···Cl1 <sup>ii</sup>	0.85 (4)	2.32 (4)	3.147 (2)	165 (3)
O1W–H2W···Cl1 <sup>iii</sup>	0.88 (3)	2.21 (3)	3.084 (2)	174 (3)
N1–H1B···Cl1	0.89	2.21	3.089 (2)	171
N1–H1C···O1W	0.89	1.81	2.697 (2)	175

Symmetry codes: (i)  $x - 1, 1 - y, z - \frac{1}{2}$ ; (ii)  $x, y, 1 + z$ ; (iii)  $x - \frac{1}{2}, \frac{3}{2} - y, \frac{1}{2} + z$ .

The aromatic and water H atoms were refined isotropically. In the aromatic ring, the refined C–H distances are in the range 0.92 (2)–0.98 (2) Å. In the water molecule, the O–H distances are 0.85 (4) and 0.88 (3) Å. The methyl and anilinium N-bound H atoms were included in the refinement in a riding-model approximation, with C–H = 0.96 Å, N–H = 0.89 Å and *U*<sub>iso</sub> = 1.5 *U*<sub>eq</sub>(carrier atom).

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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